

PAS-003-010203

Seat No.

M.Sc. (Sem. II) Examination

August/September - 2020

Physical Chemistry

Macromolecular Physical Chemistry-I
(Old Course)

Time: $2\frac{1}{2}$ Hours] [Total Marks: 70]

Instructions:

- (1) All questions are compulsory.
- (3) Each question carry 14 marks.
- 1 Answer the following: (any seven)
 - (a) Define: Regulator, copolymer, solubility parameter, telomerization.
 - (b) Write the full form and structure
 - (i) PMMA
- (ii) Teflon
- (iii) Starch
- (iv) PVA
- (c) Give an account of tacticity in polymer.
- (d) Explain cross-linking.
- (e) What is functionality? Give two examples of all types functionalities.
- (f) What is the difference between chain and stepwise polymerization?
- (g) Explain bulk polymerization.
- (h) Discuss non-linear polycondensation.
- (i) Explain polymer cyclization reaction.
- (j) What is the effect of temperature on ring scission polymerization?

Write notes on:

- (a) Discuss the reaction without rupture of main chain in thermal effect.
- (b) Discuss common initiators with suitable example.
- (c) Effect of temperature and monomer concentration on polycondensation.
- (d) State types of polymer solution and give an account of solution with $S_{\rm E} < 0$.

3 Answer the following:

- (a) Discuss kinetics of chain transfer reaction.
- (b) Discuss addition and substitution reaction.

OR

- 3 (a) Discuss entropy of mixing.
 - (b) Discuss thermodynamics of ring transfer to linear polymer.
- 4 Answer the following: (any three)
 - (a) Discuss types of solution in detail.
 - (b) Discuss molecular weight control in polycondensation.
 - (c) Discuss polyrecombination in detail.
 - (d) Explain melt and solid phase polycondensation.
- 5 Answer the following: (any two)
 - (a) Explain kinetics and mechanism of ring scission polymerization.
- ' (b) Explain:
 - (i) Solution polymerization
 - (ii) Emulsion polymerization
 - (c) Explain reactivity ratio and copolymerization behaviour.
 - (d) Discuss degradation reaction.

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